#### 1. Introduction

Aroma chemicals are an important group of organic molecules used as ingredients in flavor and fragrance composition. Aroma chemicals consist of natural, nature-identical, and artificial molecules. Natural products are obtained directly from the plant or animal sources by physical procedures. Nature-identical compounds are produced synthetically, but are chemically identical to their natural counterparts. Artificial flavor substances are compounds that have not yet been identified in plant or animal products for human consumption.

There are ca 3000 different molecules that find use in the production of flavor and fragrance compositions. Synthetic ingredients play a major part as components due to their convenient availability and the relatively lower costs compared to natural molecules from isolation of relatively limited natural sources.

#### 2. Odors Descriptors

The odors of single chemical compounds (aroma chemicals) are very difficult to describe unequivocally. The odors of complex mixtures called compounds are often impossible to describe unless one of the components is so characteristic that it determines the odor or flavor of the composition. Although an objective classification is not possible, an odor can be described by adjectives such as flowery, fruity, woody, or hay-like, which will relate to natural occurring or other well-known products with such odors characteristics.

A few terms (1) used to describe odors are listed in Table 1, with a few examples.

#### 3. General Production Routes

Aroma chemicals are specific molecules of particular aroma, which can be obtained (1) by isolation from natural sources, with or without chemical modifications, using natural molecules as precursors for many aroma chemicals (partial synthesis); (2) from petrochemical raw materials; or (3) by synthesis from cyclic and aromatic precursors.

For example, cedarwood oils obtained from plants like *Cedrus atlantica*, *Chamaecyparis funebris*, or *Juniperus mexicana*, contain aromatic molecules, eg, (E)-(+)- $\alpha$ -atlantone,  $\alpha$ -thujone, or (+)-cedrol:

$$E$$
-(+)- $\alpha$ -Atlantone  $\alpha$ -Thujone  $(+)$ -Cedrol

Table 1. Terms to Describe Odors

Odor	Description	Examples	
aldehydic	note of the long-chain fatty aldehydes, eg, fatty—sweaty, ironed laundry, seawater	n-Decanal	CHO n-Octanal
animalic	typical notes from the animal kingdom, eg, musk, castoreum, skatol, civet, ambergis	N H	
		Indole	Ambrox
balsamic	heavy, sweet odors, eg, cocoa, vanilla, cinnamon	(CH <sub>3</sub> ) <sub>2</sub> CHCO <sub>2</sub> CH <sub>3</sub> O CHO  Vanillin isobutyrate	СНО
			Cinnamaldehyde
camphor- aceous	reminiscent of camphor		OH "H
		2-Adamantanone	(+)-Isoborneol
citrus	fresh, stimulating odor of citrus fruits such as lemon or orange	СНО	СНО
earthy	humus-like, reminiscent of humid earth	Citral	Citronellal (H <sub>3</sub> C) <sub>2</sub> CHCH <sub>2</sub>
		OH 2- Ethylfenchol	6-Isobutylquinoline
fatty	reminiscent of animal fat and tallow	CH <sub>3</sub> (CH <sub>2</sub> ) <sub>8</sub> CO <sub>2</sub> (CH <sub>2</sub> ) <sub>4</sub> CH <sub>3</sub> Amyl decanoate	CHO (E,E)-2,4-Decadienal
floral, flowery	generic terms for odors of various flowers	OH Tetrahydrolinalool	CH <sub>2</sub> OH Geraniol

Table 1. (Continued)

Table 1. (	Continued)		
Odor	Description	Examples	
fruity	generic terms for odors of various fruits	CH <sub>3</sub> CO <sub>2</sub> (CH <sub>2</sub> ) <sub>7</sub> CH <sub>3</sub> ~ n-Octyl acetate	trans-2- Hexenylacetate
green	typical odor for freshly cut grass and leaves	OH cis-3-Hexenol	CHO 2-(Cyclohexyl)-propanal
herbaceous	s noncharacteristic, complex odor of green herbs with, eg, sage, minty, eucalyptus-like, or earthy nuances	Estragole	Citronellylethyl ether
medicinal	odor reminiscent of disinfectants, eg, phenol, lysol, methyl salycilate	OH Phenol	CH <sub>3</sub> O -C HO  Methyl salicylate
metallic	typical odor observed near metal surfaces, eg, brass or steel	CN 2,5-Dimethyl-2-	CH <sub>2</sub> S – SCH <sub>3</sub> Benzyl methyl
minty	peppermint-like odor	vinyl-4-hexenenitrile OH (-)-Menthol	disulfide
mossy	typical note reminiscent of forests and seaweed	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub> H <sub>3</sub> C OH OCH <sub>3</sub> Ethyl 2-hydroxy-4- methoxy-6-methyl- benzoate	OH H <sub>3</sub> C OCH <sub>3</sub> 3-Methoxy-5-methyl phenol

Table 1. (Continued)

Odor	Description	Examples	
powdery	odor identified with toilet powders, sweet-diffusive		O II CCH <sub>3</sub>
		2-(1-Cyclohexenyl)- cyclohexanone	Methyl-β-naphthyl ketone
resinous	aromatic odor of tree exudates	СНО	CH <sub>2</sub> CH <sub>2</sub> CO <sub>2</sub> H
		2-Isopropyl-5- methyl-2-hexenal	3-Phenylpropionic acid
spicy	generic term for odors of various spices	OH	O 2,4-Dimethyl-
waxy	odor resembling that of candle wax		acetophenone
		<i>n</i> -Decanal	Citronellyl-isobutyrate
woody	generic term for the odor of wood, eg, cedarwood, sandalwood	(CI	O II OCCH <sub>3</sub>
		α-Cedrene	cis-p-tert-Butyl- cyclohexylacetate

Acetylation of (+)-cedrol gives cedryl acetate, a woody-earthy odorous molecule, applied in woody compounds for all purposes.

- **3.1. The Use of Natural Molecules as Precursors.** One of the most useful sources for natural molecules as chemical precursors is turpentine oil, originated from *Pinus* sp. The oil contains 60–70% of  $\alpha$ -pinene and  $\beta$ -pinene, along with other natural molecules, ie,  $\alpha$ -phellandrene,  $\gamma$ -terpinene, anethole, caryophyllene, 3-carene, and camphene (see Figs. 1 and 2).
- **3.2.** The Use of Petrochemicals as Precursors. Synthesis from petrochemical precursors of one-to-five carbon atoms, ie, carbon monoxide/formaldehyde, acetylene, isobutylene, and isoprene, represents one of the most important routes to produce aroma chemicals.

Aromatic molecules, eg, benzene, toluene, xylenes, phenol, cresols, and naphthalene, are also important precursors for aroma chemicals (see Figs. 3–5).

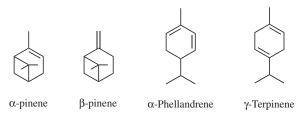
### 4. Functional Groups of Aroma Chemicals

As mentioned before, over 3000 specific chemical molecules are used in the F&F industry, but only a few hundreds are produced on a scale between 20 and 50 mt year. These molecules include most of the functional groups, from aliphatic molecules to heterocyclic ones, according to the following list:

- a. Hydrocarbons (aliphatic, acyclic terpenes, cyclic terpenes, benzenoids)
- b. Alcohols (aliphatic, alicyclic, cyclic)
- c. Ethers
- d. Aldehydes and ketones (including acetals and ketals)
- e. Carboxylic acids
- f. Esters and lactones
- g. Nitriles
- h. Amines
- i. Nitroaromatic compounds
- j. Thio compounds
- k. Heterocyclic molecules

The following sections contain selected examples of each functional group, the chemical structure, and organoleptic characteristcs.

**4.1. Hydrocarbons.** Hydrocarbons include simple aliphatic molecules, terpenes—both acyclic and cyclic, and benzene rings.



**Fig. 1.**  $\alpha$ -Pinene as a natural precursor for aroma chemicals.

Fig. 1. (Continued)

Fig. 2.  $\beta$ -Pinene as a natural precursor for aroma chemicals.

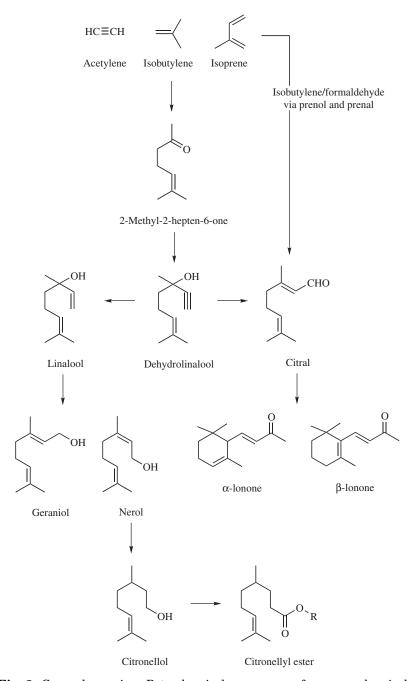


Fig. 3. General overview: Petrochemicals as a source for aroma chemicals.

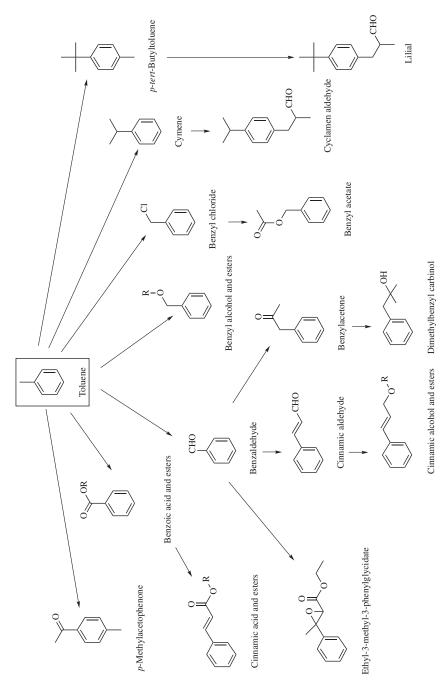


Fig. 4. Aroma chemicals derived from toluene.

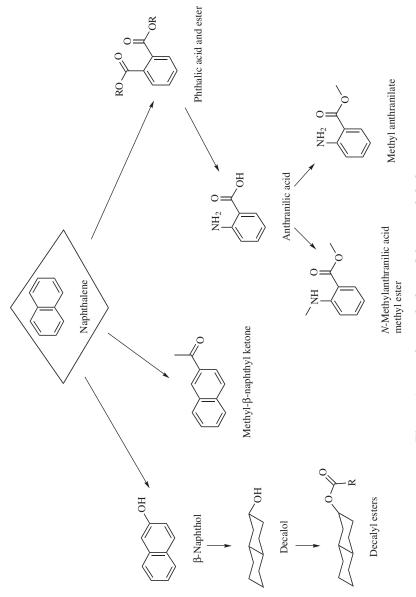
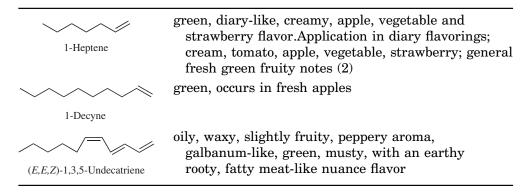


Fig. 5. Aroma chemicals derived from naphthalene.

### Unsaturated Aliphatic Non-Terpenes



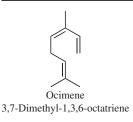
*Terpenes.* Terpenes are a group of plant originated natural products, which are usually composed of usually two, three, four, five, six or eight units of  $C_5$  atoms. These units are formally derived from 2-methyl-1,3-butadiene (isoprene).

2-Methyl-1,3-butadiene (Isoprene)

These molecules are named as follows:

Number of isoprene units	Number of carbon atoms
2	10
$rac{3}{4}$	$\begin{array}{c} 15 \\ 20 \end{array}$
5	25
6 8	30 $40$
	2 3 4

### Acyclic Monoterpenes



harsh, terpene-like, somewhat citrus; green, woody and tropical fruity



Myrcene 7-Methyl-3-methylene-1,6-octadiene harsh, terpene-like, fresh somewhat citrus in dilution

## Cyclic Monoterpenes



1-methyl-4-isopropenyl-1-cyclohexene

citrus, fruity, orange, berry-like, tarty

 $\alpha\text{-Terpinene} \\ 1\text{-methyl-4-isopropyl-1,3-cyclohexadiene}$ 

harsh, terpene-like; slightly citrus

Terpinolene 1-methyl-4-isopropylidene-1-cyclohexene

harsh, terpene-like, slightly citrus

α-Phellandrene 1-methyl-4-isopropyl-1,5-cyclohexadiene terpenic, citrus lime with a fresh green note

harsh, terpene-like, coniferous

harsh, terpene-like, coniferous

### 4.2. Bicyclic Monoterpenes



α-Pinene

2,6,6-trimethylbicyclo[3.1.1]hept-2-ene



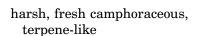
B-Pinen

 $6, 6\hbox{-}dimethyl\hbox{-}2\hbox{-}methylene bicyclo \hbox{\small [3.1.1]-heptane}$ 



Camphene

2,2-dimethyl-3-methylenebicyclo[2.2.1]-heptane



harsh, terpene-like, coniferous

citrus, herbaceous



 $\Delta$ -3-Carene

### Acyclic Sesquiterpenes



Farnesene

(3,7,11-trimethyl-1,3,6,10-dodeca-tetraene)

#### Bicyclic Sesquiterpenes

4,11,11-trimethyl-8-methylene-bicyclo-[7.2.0]undec-4-ene

spicy, woody, dusty, oily; pepper-like, camphoraceous, with a citrus background

5,6-dimethyl-8-iso-propenylbicyclo- [4.4.0]-dec-1-ene

paraffin, oily, somewhat citrus, grapefruit-like

Terpenes are formed in nature via the "two carbons metabolism", a process enabled by acetyl coenzyme A (CoA), which is produced from pyruvic acid. Acetyl-CoA forms mevalonic acid, which loses one carbon atom by decarboxylation to yield a  $C_5$  unit—isopentenyl pyrophosphate:

$$\begin{array}{c|c} O \\ CH_3C \\ -S - CoA \\ \end{array}$$

$$\begin{array}{c|c} CH_3 \\ HO_2C \\ \hline OH \\ CH_3CCH_2C \\ -S - CoA \\ \end{array}$$

$$\begin{array}{c|c} CH_3 \\ \end{array}$$

$$\begin{array}{c|c} O \\ OH \\ \end{array}$$

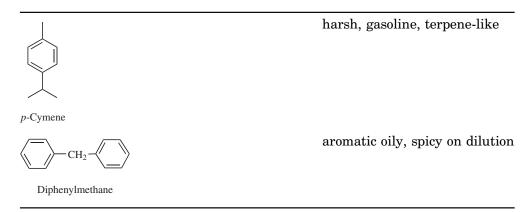
Acetoacetyl-S-Coenzyme A

Mevalonic acid Isopentenyl pyrophosphate

Two units of isopentenyl pyrophosphate are combined with one  $C_{10}$  atom unit—geranyl pyrophosphate, which loses its pyrophosphate group to form a unstabile intermediate—geranyl carbocation:

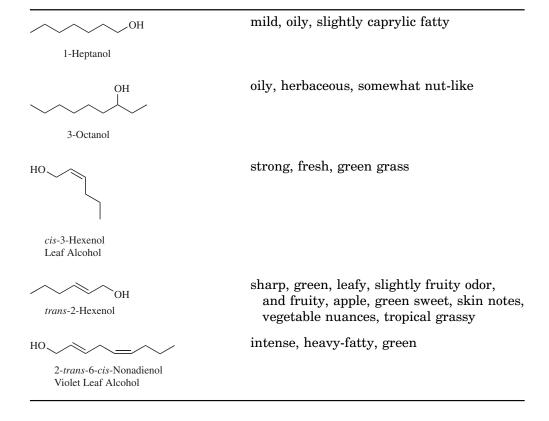
The geranyl carbocation can be stabilized by the following possibilities:

#### Benzenoids



**4.3. Alcohols.** The alcohol function is found in simple aliphatic molecules, in acyclic and cyclic terpenes, and in molecules containing benzene rings. Phenols are also contained in this group of aroma chemicals.

### Aliphatic Alcohols



### Alcohols—Acyclic Terpenes

CH<sub>2</sub>OH

Tetrahydrogeraniol 3,7-dimethyloctanol

CH<sub>2</sub>OH

Geraniol [(*E*)-isomer] 3,7-dimethyl-(*E*)-2,6-octadienol

CH<sub>2</sub>OH

Nerol [(Z)-isomer] 3,7-dimethyl-(Z)-2,6-octadienol

CH<sub>2</sub>OH

(*E,E*)-Farnesol 3,7,11-trimethyl-2,6,10-dodecatrieno

fresh floral, rosy, fatty

floral rose, citrus-like, fruity, slightly fatty

floral rose, geranium; fruity, pear; citrus-lemon

delicate, fresh green; floral (muguet)

### Alcohols—Cyclic Terpenes

ОН

Menthol (8-p-menthen-3-ol)

fresh, minty, with a dusty and earthy note

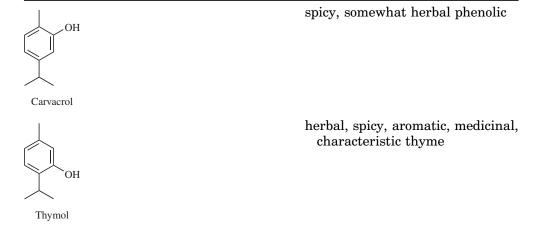
Н	natural camphoraceous, pine-needlelike
I-(-)-Borneol	
ОН	spearmint, caraway

## Alcohols Containing Benzene Rings

CH <sub>2</sub> OH	chemical, fruity with balsamic nuances
Benzyl alcohol  CH <sub>2</sub> CH <sub>2</sub> OH	mild, warm honey, fruity, sweet floral-rose
Phenethyl alcohol	

### Phenols

L-Carveol



## Preparation Methods of Alcohols

1. From Natural Sources Conversion of  $\alpha$ - and  $\beta$ -pinene to alcohols

Citronellol

Geraniol

2. From Chemical Precursors. The starting materials are isoprene, acetylene, formaldehyde, and acetone, which are used for the production of one of the possible the key intermediates for linalool and geraniol - 6-methyl-5-hepten-2-one.

6-Methyl-5-hepten-2-one

6-Methyl-5-hepten-2-one is synthesized in several routes (3):

1. From Acetylene and Acetone. Addition of acetylene to acetone, yielding 3-methyl-1-butyn-3-ol, which undergoes hydrogen addition to obtain 3-methyl-1-buten-3-ol, in presence of palladium catalyst:

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ OH & & \\ \hline \end{array} \begin{array}{c} & & \\ H_2 \\ \hline \\ OH \end{array} \begin{array}{c} & \\ \\ OH \end{array}$$

3-Methyl-1-butyn-3-ol reacts with diketene or with ethylacetoacetate as following:

2. This acetaoacetate derivative, undergoes Carroll rearrangement, accompanied by decarboxylation to give the desired product:

6-Methyl-5-hepten-2-one

3. By Claisen Rearangement. In this route, 6-methyl-5-hepten-2-one is prepared by reaction of 3-methyl-1-buten-3-ol with isopropenylmethyl ether, followed by Claisen rearangement:

4. From Acetone and Isoprene. In this route, hydrochloric acid is added to isoprene to obtain 3-methyl-2- butenylchloride. Reaction of the hydrogen chloride with acetone, in the presence of catalytic amount of organic

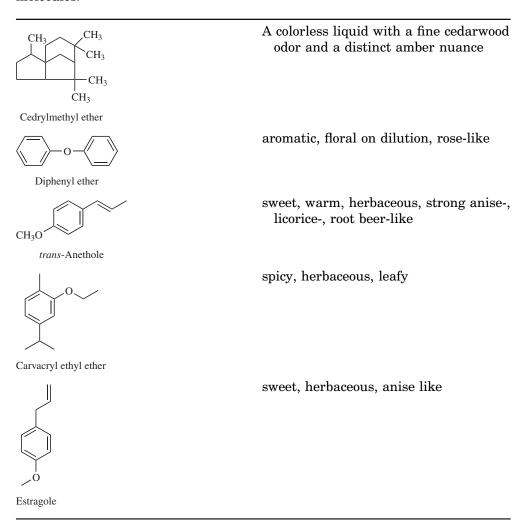
base, yields the desired product:

5. From Isobutylene and Formaldehyde. In this process 6-methyl-5-hepten-2-one is prepared via isoprenol by isomerization of 2-methyl-1-hepten-6-one. The starting material can be prepared in two steps from isobutylene and formaldehyde. The formed 3-methyl-3-buten-1-ol reacts with acetone to yield the desired product:

$$+ CH_2O \longrightarrow OH \longrightarrow OO$$

6-methyl-5-hepten-2-one, the main intermediate to linalool, can be further converted to important aroma chemicals such as geraniol, tetrahydrolinalool, methyl ionones, and others.

# **4.4. Ethers.** The ether function is found both in aliphatic and aromatic molecules.



# 4.5. Aldehydes and Ketones. Saturated Aldehydes

CHO Hexanal	aldehydic green, slightly fruity; somewhat green apple-like
Nonanal	aldehydic, peely, floral (somewhat rosy), orange
CHO Decanal	soft fatty; slightly green-fruity; cream, milk, cheese-like and green melon

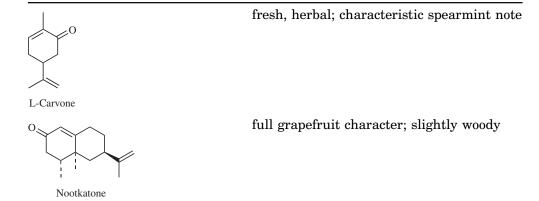
# Monounsaturated Aldehydes

CHO  trans-2-Hexenal	strongly leafy green, slightly spicy, bitter almond-like
CHO trans-2-Nonenal	green, soapy, cucumber/melon-like with an aldehydic fatty nuance
CHO 2,6-Dimethyl-5-heptenal	fresh, watery fruity (melon-like), with herbal notes.
Diunsaturated Aldehydes	
CHO  trans,trans-2,4-Decadienal	powerful fatty, aldehydic, somewhat citrus
CHC 2-trans-6-cis-Nonadienal	powerful green cucumber, melon, violet leaf; aldehydic with a fresh vegetable note
Terpene Aldehydes	
CHO	fresh lemon-like, citrus and fruity
CHO	fresh, natural, citrus, slightly fruity-herbal.
СНО	citrus, green, fruity, perfumistic, aldehydic, soapy.
Citronellal	

# Aldehydes Containing Benzene Ring

ОСНО	sweet aromatic, somewhat vanilla, characteristic heliotropic
Heliotropin	
СНО	green, herbal, spicy; characteristic cumin
Cuminaldehyde	
CH <sub>3</sub> CH <sub>2</sub> CHCHO	fresh, watery, floral, cyclamen-like
Cyclamen aldehyde	
CH <sub>3</sub> CH <sub>2</sub> CHCHO	fresh, light, green, floral, reminiscent of lily-of-the valley; notes of muguet
<b>X</b> *	
<i>p-tert</i> -Butyl-α-methyl dihydrocinnamic aldehyde	
CHO OCH <sub>3</sub>	intensive sweet, tenacious creamy, characteristic vanilla aroma
OH	
Vanillin	

### Ketones



β-Methylnaphthylketone

α-n-Methylionone

**β-Damascone** 

3-Methylcyclopentadecanone

5-Acetyl-1,1,2,3,3,6-hexamethylindan

3-Acetyl-3,4,10,10-tetramethylbicyclo [4.4.0]decane

2-Methyltetrahydrofuran-3-one

1,3,4,6,7,8-Hexahydro-4,6,6,7,8,8-hexamethylcyclopenta-(*g*)-2-benzopyran

powdery, sweet aromatic, floral; on dilution resembling neroli

floral, woody; violet-like

fruity-floral, slightly woody, herbal; somewhat raspberry connotation

natural, erogenic, animal-like musk

nitro-free musk compounds, herbal, and floral aspects

woody, amber

breadlike, buttery top-note; nutty and astringent with a slight creamy almond nuance flavor; sweet, somewhat fruity, caramellic

powerful and clean musk, approaching the aspects of macrocylic musks

Last isochormanic system drawn, 1,3,4,6,7,8-hexahydro-4,6,6,7,8,8-hexamethylcyclopenta-(g)-2-benzopyran, which was developed in the middle 1960 (3) by Beets and Heeringa from IFF is know also commercially as eg, Galaxolide, Abbalide. This molecule is synthesized as following; There is a condensation–cyclization stage of *tert*-amyl alcohol and  $\alpha$ -methyl styrene in acidic conditions to obtain the indane system, followed by a Friedel–Crafts reaction with propylene oxide to get the side chain. The side chain is finally closed to the isochromanic system using formaldehyde:

$$t$$
-Amyl alcohol  $\alpha$ -Methylstyrene 1,1,2,3,3-Pentamethylindane CH<sub>2</sub>O  $t$ -CH<sub>2</sub>OH  $t$ -CH<sub>2</sub>OH

 $\begin{array}{c} \text{CH}_2\text{CO}_2\text{CH}_3 & \text{extremel}\\ \text{charac} \\ \\ \text{C}\\ \\ \text{C}\\ \\ \text{Methyl dihydrojasmonate} \end{array}$ 

extremely persistent and powerful floral, fruity; characteristic of natural jasmin flower

indanyl-5')-propanol-1

fruity-floral, slightly woody, herbal; somewhat raspberry connotation

β-Damascone

Methyl dihydrojasmonate is known also commercially by the names Hedione, Claigeon.

Methyl dihydrojasmonate is synthesized by the following route (3) namely, Michael addition of diethyl malonate to the pentyl cyclopentenone to obtain the second side chain, followed by hydrolysis and decarboxylation, and finaly esterification:

$$(CH_2)_4 CH_3 \qquad \begin{array}{c} \xrightarrow{H_2C(COOR)_2} \\ \xrightarrow{Michael \\ addition} \end{array} \qquad \begin{array}{c} CH(COOR)_2 \\ (CH_2)_4 CH_3 \end{array}$$

2-Pentyl-2-cyclopenten-1-one

2-Pentyl-3-oxocyclopentylmalonate

Methyl dihydrojasmonate

2-Pentyl-3-oxocylopentyl acetic acid

sweet, strongly buttery, creamy, milky

#### *Diketones.* The diketones used as aroma chemicals are mostly $\alpha$ -diketones.

2,3-Butanedione (diacetyl)

creamy, sweet odor, buttery and cheesy

2,3-Hexanedione

buttery, cheesy, "oily", somewhat fruity

5-Methyl-2,3-hexanedione

"Oily", buttery, cheesy, pungent

2,3-Heptanedione

strong penetrating buttery, cheesy to slightly animal. In dilution: sweet "oily" berry note

2,3-Heptanedione

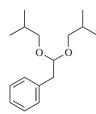
burnt caramellic flavor, and aromatic, burnt, caramelic

3,4-Hexanedione

## Acetals and Ketals

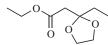
Phenylacetaldehyde dimethylacetal

dry, green-floral, fruity, citrus peel



Phenylacetaldehyde diisobutylacetal

sweet aromatic, honey, brown, somewhat floral (hyacinth-like)



Ethyl acetoacetate ethylene glycol ketal

strongly fruity, slightly floral; apple-, pear-, and berry-like

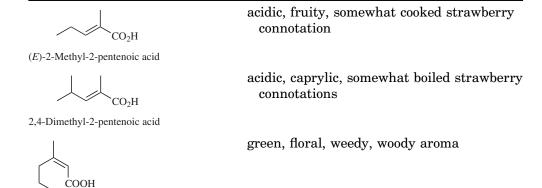
# 4.6. Carboxylic Acids. Saturated Carboxylic Acids

Table 2. Saturated Carboxylic Acids

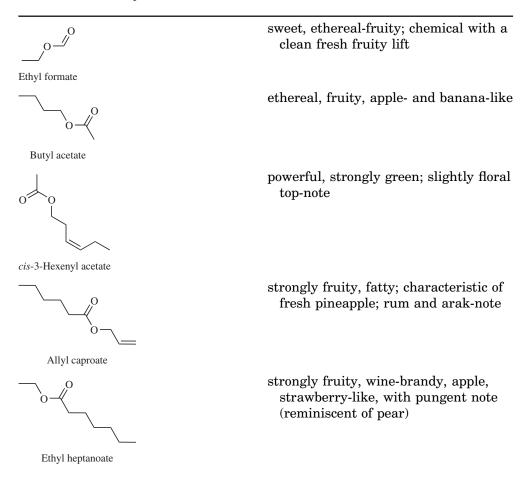
Name	Organoleptic Characteristics	structure
formic acid	pungent, acidic, sour, astringent with a fruity depth	HCO <sub>2</sub> H
acetic acid	sour, vinegar-like	$\mathrm{CH_{3}CO_{2}H}$
propionic acid	sour, fruity on dilution	$\mathrm{CH_{3}CH_{2}CO_{2}H}$
butyric acid	penetrating, reminiscent of rancid butter	$CH_3(CH_2)_2CO_2H$
valeric acid	strongly acidic, caprylic, cheese-like	$CH_3(CH_2)_3CO_2H$
caproic acid	acidic, caprylic, fatty	$CH_3(CH_2)_4CO_2H$
oenanthic acid	caprylic, fatty, green	$CH_3(CH_2)_5CO_2H$
caprylic acid	caprylic, fatty, oily	$CH_3(CH_2)_6CO_2H$
pelargonic acid	oily, fatty, caprylic; cheesy with a mild creamy background	$\mathrm{CH_{3}(CH_{2})_{7}CO_{2}H}$
capric acid	sour, fatty aroma	$CH_3(CH_2)_8CO_2H$
undecylic acid	fatty, fruity aspects	$CH_3(CH_2)_9CO_2H$
lauric acid	mild fatty	$CH_3(CH_2)_{10}CO_2H$
myristic acid	faint oily, fatty	$CH_3(CH_2)_{12}CO_2H$
palmitic acid	faint oily aroma	$CH_3(CH_2)_{14}CO_2H$
stearic acid	fatty, stearinic	$CH_3(CH_2)_{16}CO_2H$

### Unsaturated Carboxylic Acids

cis-Geranic acid



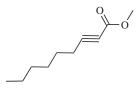
### 4.7. Carboxylic Acids Derivatives. Esters



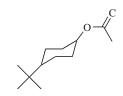
trans-2-Hexenylacetate

sweet, green, fresh, waxy and fruity; banana- and apple-like

green, violet-leaf



Methyl-2-nonynoate



sweet and rich woody, pleasant floral, with fruity note

p-tert-Butylcyclohexyl acetate

### Terpenic Esters

sweet fruity-floral, rose-, and lavender-like

Geranyl acetate

CH<sub>3</sub>CO<sub>2</sub>

freshly floral; bergamot-, petitgrain-, lavender-, and cologne-like

Linalyl acetate

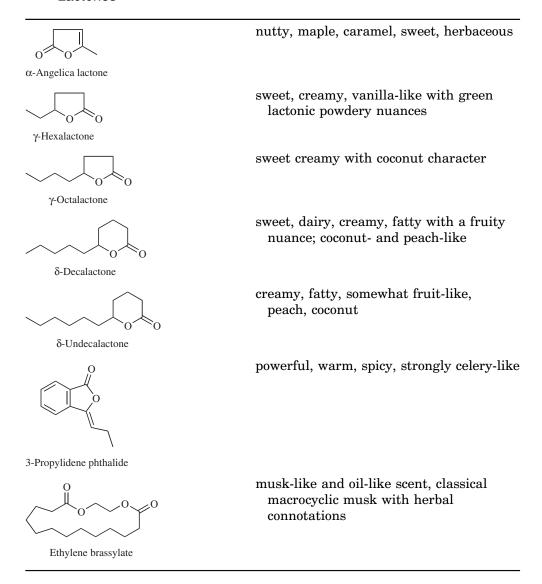
fresh, floral-herbal, slightly fruity; lavender-like

Lavandulyl acetate

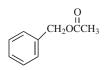
natural pineneedle-like, coniferous, camphoraceous, slightly minty

Bornyl acetate

#### Lactones



### Benzylic and Homobenzylic Esters



Benzyl acetate

green, dry-powdery, fruity, somewhat milky and estery

sweet aromatic, floral fruity, plum-like

Benzyl butyrate

honey, sweet, floral

Phenethyl acetate

### Benzoate and Homobenzoate Esters



heavy sweet, slightly floral-fruity; berry-like

Methyl benzoate

sweet aromatic, honey, waxy, fruity

Ethyl phenylacetate

#### Cinnamate Esters

fruity, balsamic, somewhat strawberry-like

Methyl cinnamate

sweet, floral, fruit, spicy; coumarin, balsamic, honey

Benzyl cinnamate

# Salicylate Esters

 0 	long lasting, green floral, leathery note
OH	
cis-3-Hexenyl-salicylate	
CH <sub>3</sub> O OCH <sub>3</sub> HO CH <sub>3</sub>	character-impact compound of oak- and treemoss; true moss-character
Methyl-3-methylorselinate	

## Nitriles

1	fresh, citrus, floral; lemon note of citral
CN	, , ,
Geranyl nitrile	
CN	fresh, lemon odor with greenish accent, citrus and herbal notes
Citronellyl nitrile	

# Amines

$\searrow$ NH <sub>2</sub>	fishy, ammonia-like; in low concentration somewhat fermented
Isopentylamine	
NH <sub>2</sub>	strong amine-like, fishy, on dilution slightly cheese like
n-Butylamine	

orangeflower-like, sweet fruity, tangerine and grape-note

Methyl anthranilate

mandarin- and grape-like, tangerine note; somewhat orange-blossom

Dimethyl anthranilate

### Nitroaromatic Compounds

	$COCH_3$		
H <sub>3</sub> C \	CH <sub>3</sub>		
$O_2N$	Y NO <sub>2</sub>		
	$C(CH_3)_3$		

dry, powdery, nitro musk; somewhat floral-fruity connotations

Musk ketone

$$\begin{array}{c|c} CH_3 & NO_2 \\ \hline \\ OCH_3 \\ C(CH_3)_3 \end{array}$$

Musk ambrette

strong nitromusk, with fruity (pear-like) note

### 4.8. Thio Compounds

S~S~S~✓

sulfurous, characteristic garlic

Diallyl trisulfide

\_s\_\_s\_

strong sulfurous, cabbage-, cauliflower- like

Dimethyl disulfide

s o

pineapple, tropical, onion at high levels

Ethyl 3-Methylthiopropionate

# **4.9. Heterocyclic Compounds.** Nonaromatic Compounds Containing Oxygen, Nitrogen, or Sulfur

fresh, strong eucalyptus-like, camphoraceous, minty, cooling

1.8-Cineole

floral, rose-like, fruity

2-Isobutenyl-4-methyltetrahydropyran (rose oxide)

sweet, caramel-fruity (pineapple-like), fried meat aspects

4-Hydroxy-2,5-dimethyl-3(2H)-furanone



sulfurous, on dilution herbal, fruity

2-Methyl-4-propyl-1,3-oxathiane

#### **Furans**

O CHO

sweet caramel-like, nutty, baked bread, almond

Furfural

O CH<sub>2</sub>SH

on dilution strong coffee-like

Furfuryl mercaptan

CH<sub>3</sub>OSCH<sub>3</sub>

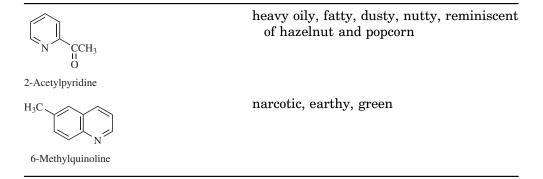
sulfurous, burnt, roasted (coffee-like) on dilution

2-Methyl-5-(methylthio)furan

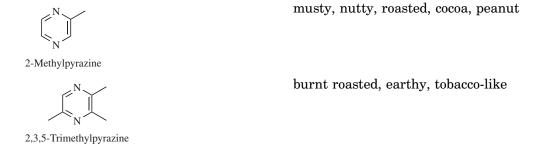
## Pyrrols and Indoles

N CCH <sub>3</sub>	sweet musty, nutty and tea-like
H II O	
N	animalic, musk, cheese, slightly fecal on dilution
Indole	
CH <sub>3</sub>	putrid, sickening, mothballs, decayed, fecal
Skatole	

## Pyridines and Quinolines



# Pyrazines and Quinoxalines



2-Acetyl-3-methylpyrazine



2-Methoxy-3-methylpyrazine



2-Methlthio-3-methyl-pyrazine



2-Methlthio-3-methyl-pyrazine



5,6,7,8-Tetrahydroquinixaline

roasted potatoes, nutty, vegetable, and cereal

roasted peanuts

nutty, sweet, weakly green

dusty, roasted

narcotic, fishy; on dilution fried and roasted aspects

#### Thiazoles

5-(2-Hydroxyethyl)-4-methylthiazole

$$\sqrt{\frac{N}{S}}$$

2-Isopropyl-4-methylthiazole

4-Methyl-5-vinylthiazole

meaty, nutty

green, vegetable character; nut-like, fruity

nutty, musty, earthy, cocoa powder like

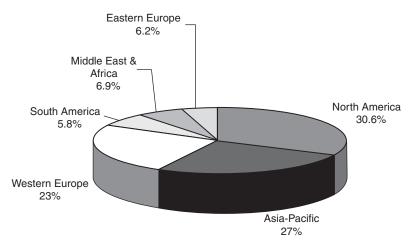


Fig. 6. Regional destribution of F&F market.

### 5. Economic Aspects

Aroma chemicals sales were expected to be  $\$2,900\times10^6$  in 2011 up from  $\$2,200\times10^6$  in 2006. Essential oils sales were expected to be  $\$\sim5,000\times10^6$  up from  $\$3,900\times10^6$  in 2006.

The regional distribution of the Flavor and Fragrance (F&F) market is shown in Fig. 6.

The top ten flavor and fragrance industry leaders and their estimated sales volume and market share are listed in Table 3.

### 6. Analytical Methods

Quality control of fragrance and flavor substances, as well as products derived from them, comprises the comparison of sensory, analytical, and it necessary, microbiological data with standards and specifications (5).

Table 3. Top Ten Fragrance and Flavor Industry Leaders<sup>a</sup>

Company	Market share, %	Sales, $\$ \times 10^6$
Givaudan	20.6	4538.4
Firmenich	13.5	2978.0
IFF	11.9	2622.9
Symise	9.6	2106.6
Takasago	6.4	1416.2
Mane SA	2.9	643.4
Sensient Flavors	$ ext{-}{\sim}2.6$	582.6
T. Hasegawa	2.5	556.9
Robertet SA	2.2	484.5
Frutarom	2.1	451.1
Total	74.5	${\sim}16381.6$
All others	25.5	${\sim}5618.4$

<sup>&</sup>lt;sup>a</sup>From Ref. 4.

In the past few decades, a precise methodology has been developed for sensory evaluation. Increasingly in recent years, chemical sensor systems have been used for this purpose.

The analytical determination of identity and purity aids greatly in establishing the acceptability of fragrance and flavor materials. To meet customer requirements, all of these methods should be validated by quality assurance tools.

Single fragrance and flavor materials are identified by generally accepted analytical parameters such as density, refractive index, optical rotation, and melting point. The advantage of these methods is the short analysis time, which provides assessment criteria allowing comparison with other laboratories around the world. Spectroscopic methods such as IR and near IR are becoming more important for fast identity checks. NIR techniques may also be used for identification of single and complex fragrance and flavor materials.

Content as well as impurity determinations are done by gas chromatography (GC), high pressure liquid chromatography (HPLC), capillary electrophoresis (CE), and by spectroscopic techniques (UV, IR, MS, and NMR GC is used for quality and in-process control to give detailed results within a few minutes (6).

Classical sample preparation methods such as distillation and Soxhlet extraction are still used, but specific techniques such as supercritical fluid extraction (SFE) and, increasingly in recent years, adsorption techniques such solid phase micro-extraction (SPME) and stir-bar extraction are also being used for isolation, separation, and identification of flavor and fragrance materials (7,8).

The determination of trace compounds such as halogens, heavy metals and pesticides, aflatoxins, residual solvents, and allergens in flavor and fragrance materials is becoming increasingly important (9–11).

For pesticides, a combination of GC-MS and LC-MS techniques is used to analyze quantities in the ppb range. Special detector systems such as electron capture detector (ECD) and atomic absorption spectroscopy (AAS) are used for detection and quantification of halogens and heavy metal content.

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